

1-[4-Chloro-2-[2-(2-fluorophenyl)-1,3-dithiolan-2-yl]phenyl]-2-methyl-1*H*-imidazole-5-carbaldehyde

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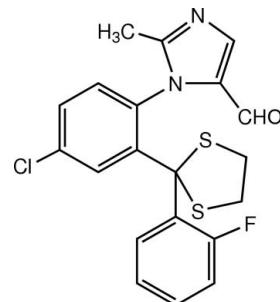
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Key indicators: single-crystal X-ray study; $T = 297\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$; disorder in main residue; R factor = 0.040; wR factor = 0.110; data-to-parameter ratio = 22.2.

There are two molecules in the asymmetric unit of the title imidazole derivative, $C_{20}H_{16}ClFN_2OS_2$. In one molecule, the dithiolane ring is disordered over two positions in a 0.849 (9):0.151 (10) ratio. The imidazole ring makes dihedral angles of 79.56 (9) and 18.45 (9) $^\circ$ with the 4-chlorophenyl and 2-fluorophenyl rings, respectively, in one molecule; in the other molecule, the corresponding angles are 82.72 (9) and 17.39 (10) $^\circ$. In the crystal, molecules are linked by weak C—H···O interactions and these linked molecules are stacked along the b axis by π – π interactions with a centroid–centroid distance of 3.4922 (11) \AA . In addition, π – π interactions between the imidazole and 2-fluorophenyl rings are also observed, with centroid–centroid distances of 3.4867 (11) and 3.4326 (10) \AA . The crystal is further consolidated by weak C—H··· π interactions. Cl···S [3.5185 (8) \AA], C···O [3.192 (3) \AA] and C···C [3.326 (2)–3.393 (3) \AA] short contacts are also observed.

Related literature

For reference bond-length data, see: Allen *et al.* (1987). For details of ring conformations, see: Cremer & Pople (1975). For background to and applications of imidazole derivatives, see: Dutta *et al.* (2009); Hori *et al.* (2000); Khabnadideh *et al.* (2003); Mamolo *et al.* (2004); Quattara *et al.* (1987); Sengupta & Bhattacharya (1983); Ucucu *et al.* (2001); Yesilada *et al.* (2004).



Experimental

Crystal data

| | |
|------------------------------|------------------------------------------|
| $C_{20}H_{16}ClFN_2OS_2$ | $V = 3797.96 (10)\text{ \AA}^3$ |
| $M_r = 418.92$ | $Z = 8$ |
| Monoclinic, $P2_1/c$ | Mo $K\alpha$ radiation |
| $a = 18.5654 (3)\text{ \AA}$ | $\mu = 0.44\text{ mm}^{-1}$ |
| $b = 9.2730 (1)\text{ \AA}$ | $T = 297\text{ K}$ |
| $c = 24.7174 (4)\text{ \AA}$ | $0.57 \times 0.52 \times 0.43\text{ mm}$ |
| $\beta = 116.807 (1)^\circ$ | |

Data collection

| | |
|-------------------------------------------------------------------|----------------------------------------|
| Bruker SMART APEXII CCD area-detector diffractometer | 42050 measured reflections |
| Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2005) | 11058 independent reflections |
| $T_{\min} = 0.787$, $T_{\max} = 0.833$ | 8238 reflections with $I > 2\sigma(I)$ |
| | $R_{\text{int}} = 0.025$ |

Refinement

| | |
|---------------------------------|-----------------------------------------------|
| $R[F^2 > 2\sigma(F^2)] = 0.040$ | 498 parameters |
| $wR(F^2) = 0.110$ | H-atom parameters constrained |
| $S = 1.01$ | $\Delta\rho_{\max} = 0.47\text{ e \AA}^{-3}$ |
| 11058 reflections | $\Delta\rho_{\min} = -0.29\text{ e \AA}^{-3}$ |

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$, $Cg2$, $Cg3$, $Cg4$ and $Cg5$ are the centroids of the $C16B$ – $C18B$ / $N1B$ / $N2B$, $C1B$ – $C6B$, $C16A$ – $C18A$ / $N1A$ / $N2A$, $C1A$ – $C6A$ and $C10A$ – $C15A$ rings, respectively.

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--------------------------------|--------------|--------------------|-------------|----------------------|
| $C17B$ – $H17B\cdots O1A$ | 0.93 | 2.27 | 3.163 (3) | 160 |
| $C20A$ – $H20C\cdots F1A$ | 0.96 | 2.53 | 3.377 (2) | 147 |
| $C2A$ – $H2AA\cdots Cg5^i$ | 0.93 | 2.81 | 3.727 (2) | 167 |
| $C13A$ – $H13A\cdots Cg4^{ii}$ | 0.93 | 2.98 | 3.7764 (19) | 144 |
| $C13B$ – $H13B\cdots Cg2^{ii}$ | 0.93 | 2.93 | 3.716 (2) | 143 |

Symmetry codes: (i) $-x + 1, y - \frac{1}{2}, -z + \frac{3}{2}$; (ii) $x, y + 1, z$.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *SAINT* (Bruker, 2005); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: WN2418).

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supplementary materials

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1-{4-Chloro-2-[2-(2-fluorophenyl)-1,3-dithiolan-2-yl]phenyl}-2-methyl-1*H*-imidazole-5-carbaldehyde

H.-K. Fun, S. Chantrapromma, V. Sumangala, G. K. Nagaraja and B. Poojary

Comment

Imidazole is a constituent part of some very important compounds, such as purine, adenine, xanthine, guanine and co-enzyme A, and many drugs contain the imidazole ring. Imidazole derivatives have occupied a unique place in the field of medicinal chemistry. They have a wide range of biological properties, such as antifungal (Hori *et al.*, 2000; Mamolo *et al.*, 2004) and antibacterial (Khambadideh *et al.*, 2003) activities. They are well known analgesic (Ucucu *et al.*, 2001), anti-inflammatory (Yesilada *et al.*, 2004), anthelmintic (Dutta *et al.*, 2009), antiparasitic (Quattara *et al.*, 1987), as well as antimicrobial (Sengupta & Bhattacharya, 1983) agents. In view of its interesting biological and pharmacological activities, the title compound was synthesized to evaluate its biological activities; its crystal structure is reported here.

Fig. 1 shows the asymmetric unit, which consists of two molecules *A* and *B* of $C_{20}H_{16}ClFN_2OS_2$. In molecule *B*, the dithiolan ring is disordered over two sites, the major component *BA* and the minor component *BB* (Fig. 1), having a refined site-occupancy ratio of 0.849 (9)/0.151 (10). In molecule *A*, the imidazole ring makes dihedral angles of 79.56 (9) and 18.45 (9) $^\circ$ with the 4-chlorophenyl and 2-fluorophenyl rings, respectively, whereas the corresponding angles in molecule *B* are 82.72 (9) and 17.39 (10) $^\circ$. The bond lengths are in normal ranges (Allen *et al.*, 1987).

The conformations of the dithiolan ring (C7–C9/S1–S2) in molecules *A* and *B* are different. In molecule *A*, the dithiolan ring is in an envelope conformation with the flap atom, C9A, 0.327 (2) Å out-of-plane, and puckering parameters $Q = 0.517$ (2) Å and $\varphi = 106.7$ (2) $^\circ$ (Cremer & Pople, 1975). In molecule *B*, the dithiolan rings of both major and minor disorder components are in half-chair conformations, with puckering parameters $Q = 0.536$ (3) Å and $\varphi = 263.5$ (3) $^\circ$ for the major component and $Q = 0.544$ (15) Å and $\varphi = 155$ (2) $^\circ$ for the minor component.

In the crystal packing (Fig. 2), the two molecules of the asymmetric unit are linked by a weak C—H \cdots O interaction (Table 1) involving the imidazole and aldehyde groups (C17B—H17B \cdots O1A), and these linked molecules are stacked along the *b* axis by π – π interactions with a $Cg1\cdots Cg3^{iii}$ distance of 3.4922 (11) Å (symmetry code; 1-x, 1-y, 2-z). In addition, π – π interactions between the imidazole and 2-fluorophenyl rings are also observed, with distances of $Cg1\cdots Cg2 = 3.4867$ (11) Å and $Cg3\cdots Cg4 = 3.4326$ (10) Å. The crystal structure is consolidated and stabilized by weak C—H \cdots π interactions (Table 1). Cl \cdots S [3.5185 (8) Å], C \cdots O [3.192 (3) Å] and C \cdots C [3.326 (2)–3.393 (3) Å] short contacts are also observed.

Experimental

To a 500 ml three-necked flask containing isopropyl alcohol (123 ml), 4-chloro-2[2-(2-fluorophenyl)-1,3-dithiolan-2-yl]-*N*-(1-aminoethylidene) benzenamine (20 g, 0.054 mole) was added, then followed by acetic acid and triethylamine (1.1 molar equivalent of each). A solution of bromomalonaldehyde (8.65 g, 0.055 mole) in 100 ml of isopropyl alcohol was added and refluxed for 6 hrs. The mixture was concentrated under vacuum at 308–313 K and the residue was treated with water (200 ml), followed by extraction with dichloromethane (100 ml). The organic layer was concentrated and the product isolated.

supplementary materials

Yellow block-shaped single crystals of the title compound suitable for *X*-ray structure determination were recrystallized from ethanol by the slow evaporation of the solvent at room temperature over a period of several days, Mp. 431–433 K.

Refinement

H atoms were positioned geometrically and allowed to ride on their parent atoms, with $d(C—H) = 0.93 \text{ \AA}$ for Csp^2 , 0.97 \AA for methylene C and 0.96 \AA for methyl C atoms. $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$, where $x = 1.5$ for methyl H and 1.2 for all other H atoms. A rotating group model was used for the methyl groups.

Figures

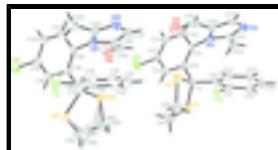


Fig. 1. The structure of the asymmetric unit of the title compound, showing 40% probability displacement ellipsoids and the atom-numbering scheme. Open bonds indicate the minor disorder component. Hydrogen atoms are shown as spheres of arbitrary radius.

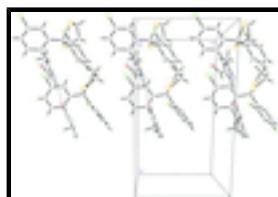


Fig. 2. The crystal packing of the major disorder component of the title compound, viewed down the c axis. Weak C—H···O interactions are shown as dashed lines.

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Crystal data

| | |
|----------------------------------|---------------------------------------------------------|
| $C_{20}H_{16}ClFN_2OS_2$ | $F(000) = 1728$ |
| $M_r = 418.92$ | $D_x = 1.465 \text{ Mg m}^{-3}$ |
| Monoclinic, $P2_1/c$ | Melting point = 431–433 K |
| Hall symbol: -P 2ybc | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| $a = 18.5654 (3) \text{ \AA}$ | Cell parameters from 11058 reflections |
| $b = 9.2730 (1) \text{ \AA}$ | $\theta = 2.3\text{--}30.0^\circ$ |
| $c = 24.7174 (4) \text{ \AA}$ | $\mu = 0.44 \text{ mm}^{-1}$ |
| $\beta = 116.807 (1)^\circ$ | $T = 297 \text{ K}$ |
| $V = 3797.96 (10) \text{ \AA}^3$ | Block, yellow |
| $Z = 8$ | $0.57 \times 0.52 \times 0.43 \text{ mm}$ |

Data collection

| | |
|-------------------------------------------------------------------|------------------------------------------------------------------------|
| Bruker SMART APEXII CCD area-detector diffractometer | 11058 independent reflections |
| Radiation source: sealed tube graphite | 8238 reflections with $I > 2\sigma(I)$ |
| φ and ω scans | $R_{\text{int}} = 0.025$ |
| Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2005) | $\theta_{\text{max}} = 30.0^\circ$, $\theta_{\text{min}} = 2.3^\circ$ |
| | $h = -25\text{--}26$ |

$T_{\min} = 0.787$, $T_{\max} = 0.833$
42050 measured reflections

$k = -11 \rightarrow 13$
 $l = -34 \rightarrow 34$

Refinement

| | |
|---------------------------------|-------------------------------------------------------------------------------------|
| Refinement on F^2 | Primary atom site location: structure-invariant direct methods |
| Least-squares matrix: full | Secondary atom site location: difference Fourier map |
| $R[F^2 > 2\sigma(F^2)] = 0.040$ | Hydrogen site location: inferred from neighbouring sites |
| $wR(F^2) = 0.110$ | H-atom parameters constrained |
| $S = 1.01$ | $w = 1/[\sigma^2(F_o^2) + (0.0494P)^2 + 1.2334P]$ where $P = (F_o^2 + 2F_c^2)/3$ |
| 11058 reflections | $(\Delta/\sigma)_{\max} = 0.001$ |
| 498 parameters | $\Delta\rho_{\max} = 0.47 \text{ e \AA}^{-3}$ |
| 0 restraints | $\Delta\rho_{\min} = -0.29 \text{ e \AA}^{-3}$ |

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | $U_{\text{iso}}^*/U_{\text{eq}}$ | Occ. (<1) |
|------|--------------|---------------|--------------|----------------------------------|-----------|
| Cl1A | 0.25942 (3) | 0.62718 (5) | 0.73463 (3) | 0.06226 (14) | |
| S1A | 0.30891 (3) | 0.06420 (5) | 0.76209 (2) | 0.04776 (11) | |
| S2A | 0.35008 (3) | 0.12181 (5) | 0.89061 (2) | 0.04827 (11) | |
| F1A | 0.45561 (7) | 0.10894 (12) | 0.75343 (5) | 0.0543 (3) | |
| O1A | 0.45746 (10) | 0.3552 (2) | 0.97799 (8) | 0.0808 (5) | |
| N1A | 0.54863 (7) | 0.30739 (13) | 0.90641 (6) | 0.0333 (3) | |
| N2A | 0.67475 (8) | 0.22391 (16) | 0.94763 (7) | 0.0462 (3) | |
| C1A | 0.49232 (10) | 0.03084 (17) | 0.80578 (8) | 0.0411 (3) | |
| C2A | 0.55553 (11) | -0.0572 (2) | 0.81286 (10) | 0.0536 (5) | |
| H2AA | 0.5715 | -0.0656 | 0.7823 | 0.064* | |
| C3A | 0.59483 (12) | -0.1329 (2) | 0.86655 (11) | 0.0603 (5) | |
| H3AA | 0.6381 | -0.1927 | 0.8727 | 0.072* | |
| C4A | 0.56967 (12) | -0.1193 (2) | 0.91058 (10) | 0.0562 (5) | |
| H4AA | 0.5967 | -0.1692 | 0.9469 | 0.067* | |
| C5A | 0.50453 (10) | -0.03238 (17) | 0.90177 (8) | 0.0439 (4) | |
| H5AA | 0.4878 | -0.0263 | 0.9319 | 0.053* | |

supplementary materials

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|------|---------------|--------------|--------------|--------------|
| C6A | 0.46387 (9) | 0.04589 (15) | 0.84846 (7) | 0.0352 (3) |
| C7A | 0.38948 (9) | 0.13927 (16) | 0.83468 (7) | 0.0342 (3) |
| C8A | 0.23702 (12) | 0.0129 (3) | 0.78845 (10) | 0.0632 (5) |
| H8AA | 0.2005 | 0.0923 | 0.7835 | 0.076* |
| H8AB | 0.2055 | -0.0688 | 0.7654 | 0.076* |
| C9A | 0.28267 (13) | -0.0262 (2) | 0.85399 (11) | 0.0617 (5) |
| H9AA | 0.3130 | -0.1145 | 0.8587 | 0.074* |
| H9AB | 0.2459 | -0.0405 | 0.8716 | 0.074* |
| C10A | 0.40297 (9) | 0.30204 (16) | 0.82977 (7) | 0.0328 (3) |
| C11A | 0.33688 (9) | 0.38578 (18) | 0.79124 (8) | 0.0391 (3) |
| H11A | 0.2872 | 0.3414 | 0.7689 | 0.047* |
| C12A | 0.34365 (9) | 0.53227 (18) | 0.78570 (8) | 0.0403 (3) |
| C13A | 0.41532 (10) | 0.60404 (18) | 0.81777 (8) | 0.0449 (4) |
| H13A | 0.4193 | 0.7029 | 0.8135 | 0.054* |
| C14A | 0.48118 (10) | 0.52407 (17) | 0.85662 (8) | 0.0418 (4) |
| H14A | 0.5303 | 0.5702 | 0.8788 | 0.050* |
| C15A | 0.47569 (9) | 0.37588 (16) | 0.86333 (7) | 0.0330 (3) |
| C16A | 0.61461 (9) | 0.27685 (17) | 0.89769 (7) | 0.0368 (3) |
| C17A | 0.64670 (11) | 0.22370 (19) | 0.98946 (8) | 0.0463 (4) |
| H17A | 0.6764 | 0.1917 | 1.0291 | 0.056* |
| C18A | 0.56970 (10) | 0.27589 (17) | 0.96678 (7) | 0.0392 (3) |
| C19A | 0.52354 (13) | 0.3053 (2) | 0.99857 (9) | 0.0527 (4) |
| H19A | 0.5476 | 0.2831 | 1.0397 | 0.063* |
| C20A | 0.61982 (11) | 0.3074 (2) | 0.84079 (8) | 0.0501 (4) |
| H20A | 0.6601 | 0.2465 | 0.8384 | 0.075* |
| H20B | 0.6342 | 0.4066 | 0.8403 | 0.075* |
| H20C | 0.5685 | 0.2887 | 0.8068 | 0.075* |
| Cl1B | -0.11743 (3) | 0.82366 (6) | 0.82366 (2) | 0.06218 (14) |
| S1B | -0.06051 (3) | 0.25797 (5) | 0.84138 (2) | 0.04791 (11) |
| S2B | 0.07738 (3) | 0.33592 (5) | 0.81837 (2) | 0.04785 (11) |
| F1B | 0.00272 (8) | 0.29431 (15) | 0.97221 (5) | 0.0681 (3) |
| O1B | 0.20278 (10) | 0.6108 (2) | 0.86047 (7) | 0.0772 (5) |
| N1B | 0.18622 (7) | 0.50963 (14) | 0.96952 (6) | 0.0348 (3) |
| N2B | 0.28898 (9) | 0.41459 (16) | 1.04951 (7) | 0.0475 (3) |
| C1B | 0.06791 (12) | 0.2203 (2) | 0.97587 (8) | 0.0479 (4) |
| C2B | 0.10655 (16) | 0.1275 (2) | 1.02414 (9) | 0.0659 (6) |
| H2BA | 0.0872 | 0.1141 | 1.0525 | 0.079* |
| C3B | 0.17401 (15) | 0.0555 (2) | 1.02951 (10) | 0.0688 (6) |
| H3BA | 0.2010 | -0.0071 | 1.0618 | 0.083* |
| C4B | 0.20137 (12) | 0.0759 (2) | 0.98734 (10) | 0.0593 (5) |
| H4BA | 0.2476 | 0.0281 | 0.9913 | 0.071* |
| C5B | 0.16075 (10) | 0.16754 (18) | 0.93844 (8) | 0.0448 (4) |
| H5BA | 0.1797 | 0.1786 | 0.9097 | 0.054* |
| C6B | 0.09254 (9) | 0.24306 (16) | 0.93159 (7) | 0.0357 (3) |
| C7B | 0.04282 (9) | 0.33975 (17) | 0.87717 (7) | 0.0336 (3) |
| C10B | 0.04177 (9) | 0.50073 (16) | 0.89252 (6) | 0.0321 (3) |
| C11B | -0.02824 (9) | 0.58123 (18) | 0.86003 (7) | 0.0374 (3) |
| H11B | -0.0750 | 0.5347 | 0.8329 | 0.045* |
| C12B | -0.02917 (10) | 0.72888 (18) | 0.86748 (7) | 0.0401 (3) |

| | | | | | |
|------|--------------|--------------|--------------|-------------|------------|
| C13B | 0.03757 (11) | 0.80291 (19) | 0.90805 (8) | 0.0464 (4) | |
| H13B | 0.0361 | 0.9022 | 0.9128 | 0.056* | |
| C14B | 0.10674 (10) | 0.72464 (18) | 0.94136 (8) | 0.0445 (4) | |
| H14B | 0.1525 | 0.7718 | 0.9696 | 0.053* | |
| C15B | 0.10945 (9) | 0.57683 (17) | 0.93361 (7) | 0.0342 (3) | |
| C16B | 0.21586 (10) | 0.47013 (18) | 1.02890 (7) | 0.0399 (3) | |
| C17B | 0.30732 (10) | 0.42030 (19) | 1.00227 (9) | 0.0468 (4) | |
| H17B | 0.3557 | 0.3880 | 1.0039 | 0.056* | |
| C18B | 0.24570 (9) | 0.47949 (17) | 0.95176 (8) | 0.0396 (3) | |
| C19B | 0.24897 (12) | 0.5274 (2) | 0.89772 (9) | 0.0532 (4) | |
| H19B | 0.2906 | 0.4909 | 0.8906 | 0.064* | |
| C20B | 0.17221 (12) | 0.4943 (3) | 1.06578 (9) | 0.0579 (5) | |
| H20D | 0.1919 | 0.4284 | 1.0993 | 0.087* | |
| H20E | 0.1155 | 0.4786 | 1.0411 | 0.087* | |
| H20F | 0.1809 | 0.5915 | 1.0807 | 0.087* | |
| C8BA | -0.0620 (2) | 0.1950 (5) | 0.77250 (17) | 0.0664 (10) | 0.849 (10) |
| H8BA | -0.0914 | 0.1048 | 0.7604 | 0.080* | 0.849 (10) |
| H8BB | -0.0889 | 0.2650 | 0.7404 | 0.080* | 0.849 (10) |
| C9BA | 0.0235 (2) | 0.1734 (4) | 0.78267 (16) | 0.0593 (9) | 0.849 (10) |
| H9BA | 0.0250 | 0.1576 | 0.7444 | 0.071* | 0.849 (10) |
| H9BB | 0.0474 | 0.0907 | 0.8086 | 0.071* | 0.849 (10) |
| C8BB | -0.0388 (12) | 0.1405 (19) | 0.7960 (9) | 0.053 (4)* | 0.151 (10) |
| H8BC | -0.0007 | 0.0679 | 0.8208 | 0.064* | 0.151 (10) |
| H8BD | -0.0877 | 0.0923 | 0.7676 | 0.064* | 0.151 (10) |
| C9BB | -0.0027 (15) | 0.228 (3) | 0.7614 (11) | 0.075 (6)* | 0.151 (10) |
| H9BC | -0.0434 | 0.2900 | 0.7314 | 0.090* | 0.151 (10) |
| H9BD | 0.0188 | 0.1649 | 0.7410 | 0.090* | 0.151 (10) |

Atomic displacement parameters (\AA^2)

| | U^{11} | U^{22} | U^{33} | U^{12} | U^{13} | U^{23} |
|------|-------------|-------------|-------------|---------------|--------------|---------------|
| Cl1A | 0.0383 (2) | 0.0507 (3) | 0.0845 (4) | 0.0141 (2) | 0.0160 (2) | 0.0254 (2) |
| S1A | 0.0388 (2) | 0.0450 (2) | 0.0457 (2) | -0.00736 (18) | 0.00682 (18) | -0.00719 (18) |
| S2A | 0.0500 (3) | 0.0469 (2) | 0.0538 (3) | -0.0069 (2) | 0.0286 (2) | -0.0025 (2) |
| F1A | 0.0612 (7) | 0.0563 (6) | 0.0475 (6) | 0.0087 (5) | 0.0264 (5) | 0.0081 (5) |
| O1A | 0.0655 (10) | 0.1031 (14) | 0.0858 (12) | -0.0056 (10) | 0.0447 (9) | -0.0174 (10) |
| N1A | 0.0274 (6) | 0.0310 (6) | 0.0355 (6) | -0.0008 (5) | 0.0090 (5) | -0.0007 (5) |
| N2A | 0.0352 (7) | 0.0425 (8) | 0.0498 (8) | 0.0069 (6) | 0.0093 (6) | 0.0016 (6) |
| C1A | 0.0399 (8) | 0.0325 (8) | 0.0464 (9) | -0.0002 (7) | 0.0155 (7) | -0.0011 (7) |
| C2A | 0.0495 (10) | 0.0407 (10) | 0.0730 (13) | 0.0015 (8) | 0.0299 (10) | -0.0093 (9) |
| C3A | 0.0413 (10) | 0.0355 (9) | 0.0924 (16) | 0.0082 (8) | 0.0199 (10) | -0.0023 (10) |
| C4A | 0.0462 (10) | 0.0342 (9) | 0.0649 (12) | 0.0039 (8) | 0.0046 (9) | 0.0097 (8) |
| C5A | 0.0442 (9) | 0.0304 (8) | 0.0469 (9) | -0.0026 (7) | 0.0116 (8) | 0.0054 (7) |
| C6A | 0.0329 (7) | 0.0236 (7) | 0.0421 (8) | -0.0023 (6) | 0.0107 (6) | -0.0008 (6) |
| C7A | 0.0303 (7) | 0.0304 (7) | 0.0369 (7) | -0.0023 (6) | 0.0106 (6) | 0.0005 (6) |
| C8A | 0.0393 (10) | 0.0584 (12) | 0.0797 (15) | -0.0132 (9) | 0.0161 (10) | -0.0025 (11) |
| C9A | 0.0541 (12) | 0.0507 (11) | 0.0834 (15) | -0.0127 (9) | 0.0336 (11) | 0.0067 (10) |
| C10A | 0.0286 (7) | 0.0283 (7) | 0.0390 (7) | 0.0003 (6) | 0.0132 (6) | 0.0005 (6) |

supplementary materials

| | | | | | | |
|------|-------------|-------------|-------------|---------------|--------------|---------------|
| C11A | 0.0260 (7) | 0.0371 (8) | 0.0484 (9) | 0.0002 (6) | 0.0116 (7) | 0.0037 (7) |
| C12A | 0.0310 (8) | 0.0368 (8) | 0.0495 (9) | 0.0078 (6) | 0.0148 (7) | 0.0082 (7) |
| C13A | 0.0444 (9) | 0.0281 (8) | 0.0575 (10) | 0.0018 (7) | 0.0190 (8) | 0.0032 (7) |
| C14A | 0.0332 (8) | 0.0323 (8) | 0.0506 (9) | -0.0043 (6) | 0.0106 (7) | -0.0018 (7) |
| C15A | 0.0272 (7) | 0.0301 (7) | 0.0374 (7) | 0.0024 (6) | 0.0107 (6) | -0.0002 (6) |
| C16A | 0.0307 (7) | 0.0320 (7) | 0.0420 (8) | 0.0007 (6) | 0.0112 (6) | -0.0025 (6) |
| C17A | 0.0457 (9) | 0.0394 (9) | 0.0389 (8) | 0.0040 (7) | 0.0060 (7) | 0.0040 (7) |
| C18A | 0.0420 (9) | 0.0329 (8) | 0.0373 (8) | -0.0035 (7) | 0.0132 (7) | -0.0004 (6) |
| C19A | 0.0556 (12) | 0.0522 (11) | 0.0548 (11) | -0.0113 (9) | 0.0287 (10) | -0.0083 (9) |
| C20A | 0.0441 (10) | 0.0569 (11) | 0.0514 (10) | -0.0008 (8) | 0.0234 (8) | -0.0001 (8) |
| C11B | 0.0466 (3) | 0.0566 (3) | 0.0570 (3) | 0.0220 (2) | 0.0001 (2) | 0.0008 (2) |
| S1B | 0.0339 (2) | 0.0497 (3) | 0.0522 (2) | -0.00964 (18) | 0.01240 (19) | -0.0126 (2) |
| S2B | 0.0566 (3) | 0.0502 (3) | 0.0435 (2) | -0.0079 (2) | 0.0286 (2) | -0.00982 (19) |
| F1B | 0.0733 (8) | 0.0838 (9) | 0.0640 (7) | 0.0218 (7) | 0.0458 (7) | 0.0078 (6) |
| O1B | 0.0712 (10) | 0.0979 (13) | 0.0585 (9) | 0.0014 (9) | 0.0257 (8) | 0.0205 (9) |
| N1B | 0.0249 (6) | 0.0350 (7) | 0.0365 (6) | 0.0012 (5) | 0.0069 (5) | -0.0032 (5) |
| N2B | 0.0362 (7) | 0.0430 (8) | 0.0487 (8) | 0.0058 (6) | 0.0064 (6) | 0.0042 (6) |
| C1B | 0.0539 (11) | 0.0463 (10) | 0.0429 (9) | 0.0055 (8) | 0.0213 (8) | 0.0000 (8) |
| C2B | 0.0898 (17) | 0.0588 (13) | 0.0454 (10) | 0.0042 (12) | 0.0271 (11) | 0.0068 (9) |
| C3B | 0.0797 (15) | 0.0440 (11) | 0.0531 (12) | 0.0071 (11) | 0.0038 (11) | 0.0071 (9) |
| C4B | 0.0476 (11) | 0.0357 (9) | 0.0721 (13) | 0.0082 (8) | 0.0072 (10) | -0.0021 (9) |
| C5B | 0.0380 (8) | 0.0334 (8) | 0.0568 (10) | 0.0018 (7) | 0.0160 (8) | -0.0068 (7) |
| C6B | 0.0359 (8) | 0.0299 (7) | 0.0364 (7) | 0.0001 (6) | 0.0119 (6) | -0.0053 (6) |
| C7B | 0.0295 (7) | 0.0355 (8) | 0.0321 (7) | -0.0017 (6) | 0.0106 (6) | -0.0063 (6) |
| C10B | 0.0295 (7) | 0.0340 (8) | 0.0296 (7) | 0.0009 (6) | 0.0107 (6) | -0.0023 (6) |
| C11B | 0.0293 (7) | 0.0421 (8) | 0.0316 (7) | 0.0022 (6) | 0.0055 (6) | -0.0038 (6) |
| C12B | 0.0332 (8) | 0.0421 (9) | 0.0358 (8) | 0.0102 (7) | 0.0075 (6) | 0.0016 (7) |
| C13B | 0.0429 (9) | 0.0335 (8) | 0.0520 (10) | 0.0047 (7) | 0.0120 (8) | -0.0036 (7) |
| C14B | 0.0343 (8) | 0.0374 (8) | 0.0490 (9) | -0.0004 (7) | 0.0074 (7) | -0.0078 (7) |
| C15B | 0.0262 (7) | 0.0358 (8) | 0.0340 (7) | 0.0023 (6) | 0.0077 (6) | -0.0023 (6) |
| C16B | 0.0342 (8) | 0.0367 (8) | 0.0378 (8) | -0.0001 (6) | 0.0065 (7) | -0.0031 (6) |
| C17B | 0.0311 (8) | 0.0385 (9) | 0.0629 (11) | 0.0052 (7) | 0.0142 (8) | 0.0012 (8) |
| C18B | 0.0309 (7) | 0.0342 (8) | 0.0508 (9) | -0.0008 (6) | 0.0159 (7) | -0.0034 (7) |
| C19B | 0.0470 (10) | 0.0601 (12) | 0.0551 (11) | -0.0074 (9) | 0.0253 (9) | -0.0050 (9) |
| C20B | 0.0550 (11) | 0.0702 (13) | 0.0440 (10) | 0.0043 (10) | 0.0184 (9) | -0.0057 (9) |
| C8BA | 0.0594 (17) | 0.074 (2) | 0.0480 (16) | -0.0147 (15) | 0.0090 (14) | -0.0239 (16) |
| C9BA | 0.0722 (19) | 0.0551 (17) | 0.0508 (16) | -0.0098 (14) | 0.0279 (15) | -0.0260 (13) |

Geometric parameters (\AA , $^\circ$)

| | | | |
|----------|-------------|----------|-------------|
| C1A—C12A | 1.7415 (16) | S2B—C9BA | 1.804 (3) |
| S1A—C8A | 1.794 (2) | S2B—C9BB | 1.82 (2) |
| S1A—C7A | 1.8764 (16) | S2B—C7B | 1.8349 (15) |
| S2A—C9A | 1.802 (2) | F1B—C1B | 1.358 (2) |
| S2A—C7A | 1.8396 (16) | O1B—C19B | 1.212 (3) |
| F1A—C1A | 1.367 (2) | N1B—C16B | 1.365 (2) |
| O1A—C19A | 1.190 (2) | N1B—C18B | 1.388 (2) |
| N1A—C16A | 1.3659 (19) | N1B—C15B | 1.4375 (19) |
| N1A—C18A | 1.392 (2) | N2B—C16B | 1.321 (2) |

| | | | |
|---------------|-------------|---------------|-------------|
| N1A—C15A | 1.4389 (19) | N2B—C17B | 1.357 (2) |
| N2A—C16A | 1.330 (2) | C1B—C6B | 1.379 (2) |
| N2A—C17A | 1.351 (2) | C1B—C2B | 1.381 (3) |
| C1A—C2A | 1.374 (2) | C2B—C3B | 1.372 (3) |
| C1A—C6A | 1.383 (2) | C2B—H2BA | 0.9300 |
| C2A—C3A | 1.383 (3) | C3B—C4B | 1.363 (3) |
| C2A—H2AA | 0.9300 | C3B—H3BA | 0.9300 |
| C3A—C4A | 1.370 (3) | C4B—C5B | 1.389 (3) |
| C3A—H3AA | 0.9300 | C4B—H4BA | 0.9300 |
| C4A—C5A | 1.387 (3) | C5B—C6B | 1.389 (2) |
| C4A—H4AA | 0.9300 | C5B—H5BA | 0.9300 |
| C5A—C6A | 1.391 (2) | C6B—C7B | 1.531 (2) |
| C5A—H5AA | 0.9300 | C7B—C10B | 1.542 (2) |
| C6A—C7A | 1.533 (2) | C10B—C11B | 1.397 (2) |
| C7A—C10A | 1.544 (2) | C10B—C15B | 1.399 (2) |
| C8A—C9A | 1.495 (3) | C11B—C12B | 1.383 (2) |
| C8A—H8AA | 0.9700 | C11B—H11B | 0.9300 |
| C8A—H8AB | 0.9700 | C12B—C13B | 1.375 (2) |
| C9A—H9AA | 0.9700 | C13B—C14B | 1.378 (2) |
| C9A—H9AB | 0.9700 | C13B—H13B | 0.9300 |
| C10A—C11A | 1.400 (2) | C14B—C15B | 1.388 (2) |
| C10A—C15A | 1.403 (2) | C14B—H14B | 0.9300 |
| C11A—C12A | 1.377 (2) | C16B—C20B | 1.485 (2) |
| C11A—H11A | 0.9300 | C17B—C18B | 1.371 (2) |
| C12A—C13A | 1.375 (2) | C17B—H17B | 0.9300 |
| C13A—C14A | 1.381 (2) | C18B—C19B | 1.435 (3) |
| C13A—H13A | 0.9300 | C19B—H19B | 0.9300 |
| C14A—C15A | 1.393 (2) | C20B—H20D | 0.9600 |
| C14A—H14A | 0.9300 | C20B—H20E | 0.9600 |
| C16A—C20A | 1.480 (2) | C20B—H20F | 0.9600 |
| C17A—C18A | 1.367 (2) | C8BA—C9BA | 1.504 (5) |
| C17A—H17A | 0.9300 | C8BA—H8BA | 0.9700 |
| C18A—C19A | 1.426 (2) | C8BA—H8BB | 0.9700 |
| C19A—H19A | 0.9300 | C9BA—H9BA | 0.9700 |
| C20A—H20A | 0.9600 | C9BA—H9BB | 0.9700 |
| C20A—H20B | 0.9600 | C8BB—C9BB | 1.54 (3) |
| C20A—H20C | 0.9600 | C8BB—H8BC | 0.9700 |
| C11B—C12B | 1.7407 (16) | C8BB—H8BD | 0.9700 |
| S1B—C8BB | 1.737 (15) | C9BB—H9BC | 0.9700 |
| S1B—C8BA | 1.788 (3) | C9BB—H9BD | 0.9700 |
| S1B—C7B | 1.8728 (15) | | |
| C8A—S1A—C7A | 98.85 (9) | C16B—N2B—C17B | 105.43 (14) |
| C9A—S2A—C7A | 96.37 (9) | F1B—C1B—C6B | 118.32 (15) |
| C16A—N1A—C18A | 106.68 (13) | F1B—C1B—C2B | 118.18 (18) |
| C16A—N1A—C15A | 125.89 (13) | C6B—C1B—C2B | 123.49 (18) |
| C18A—N1A—C15A | 126.99 (13) | C3B—C2B—C1B | 118.9 (2) |
| C16A—N2A—C17A | 105.00 (14) | C3B—C2B—H2BA | 120.6 |
| F1A—C1A—C2A | 117.88 (16) | C1B—C2B—H2BA | 120.6 |
| F1A—C1A—C6A | 118.02 (14) | C4B—C3B—C2B | 119.8 (2) |

supplementary materials

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|----------------|-------------|----------------|-------------|
| C2A—C1A—C6A | 124.10 (17) | C4B—C3B—H3BA | 120.1 |
| C1A—C2A—C3A | 118.32 (19) | C2B—C3B—H3BA | 120.1 |
| C1A—C2A—H2AA | 120.8 | C3B—C4B—C5B | 120.48 (19) |
| C3A—C2A—H2AA | 120.8 | C3B—C4B—H4BA | 119.8 |
| C4A—C3A—C2A | 119.60 (17) | C5B—C4B—H4BA | 119.8 |
| C4A—C3A—H3AA | 120.2 | C6B—C5B—C4B | 121.45 (19) |
| C2A—C3A—H3AA | 120.2 | C6B—C5B—H5BA | 119.3 |
| C3A—C4A—C5A | 120.96 (18) | C4B—C5B—H5BA | 119.3 |
| C3A—C4A—H4AA | 119.5 | C1B—C6B—C5B | 115.88 (16) |
| C5A—C4A—H4AA | 119.5 | C1B—C6B—C7B | 120.13 (14) |
| C4A—C5A—C6A | 120.95 (18) | C5B—C6B—C7B | 123.89 (15) |
| C4A—C5A—H5AA | 119.5 | C6B—C7B—C10B | 114.84 (12) |
| C6A—C5A—H5AA | 119.5 | C6B—C7B—S2B | 113.37 (10) |
| C1A—C6A—C5A | 116.03 (15) | C10B—C7B—S2B | 105.14 (10) |
| C1A—C6A—C7A | 120.22 (14) | C6B—C7B—S1B | 105.13 (10) |
| C5A—C6A—C7A | 123.69 (15) | C10B—C7B—S1B | 112.60 (10) |
| C6A—C7A—C10A | 114.07 (12) | S2B—C7B—S1B | 105.52 (7) |
| C6A—C7A—S2A | 114.03 (11) | C11B—C10B—C15B | 116.49 (14) |
| C10A—C7A—S2A | 106.19 (10) | C11B—C10B—C7B | 119.18 (13) |
| C6A—C7A—S1A | 104.64 (10) | C15B—C10B—C7B | 124.05 (13) |
| C10A—C7A—S1A | 112.03 (10) | C12B—C11B—C10B | 121.13 (15) |
| S2A—C7A—S1A | 105.68 (7) | C12B—C11B—H11B | 119.4 |
| C9A—C8A—S1A | 107.87 (13) | C10B—C11B—H11B | 119.4 |
| C9A—C8A—H8AA | 110.1 | C13B—C12B—C11B | 122.02 (15) |
| S1A—C8A—H8AA | 110.1 | C13B—C12B—Cl1B | 119.17 (13) |
| C9A—C8A—H8AB | 110.1 | C11B—C12B—Cl1B | 118.81 (13) |
| S1A—C8A—H8AB | 110.1 | C12B—C13B—C14B | 117.58 (16) |
| H8AA—C8A—H8AB | 108.4 | C12B—C13B—H13B | 121.2 |
| C8A—C9A—S2A | 106.58 (14) | C14B—C13B—H13B | 121.2 |
| C8A—C9A—H9AA | 110.4 | C13B—C14B—C15B | 121.29 (15) |
| S2A—C9A—H9AA | 110.4 | C13B—C14B—H14B | 119.4 |
| C8A—C9A—H9AB | 110.4 | C15B—C14B—H14B | 119.4 |
| S2A—C9A—H9AB | 110.4 | C14B—C15B—C10B | 121.45 (14) |
| H9AA—C9A—H9AB | 108.6 | C14B—C15B—N1B | 115.49 (14) |
| C11A—C10A—C15A | 116.47 (14) | C10B—C15B—N1B | 123.05 (14) |
| C11A—C10A—C7A | 118.17 (13) | N2B—C16B—N1B | 111.43 (15) |
| C15A—C10A—C7A | 125.28 (13) | N2B—C16B—C20B | 124.95 (16) |
| C12A—C11A—C10A | 121.56 (15) | N1B—C16B—C20B | 123.55 (15) |
| C12A—C11A—H11A | 119.2 | N2B—C17B—C18B | 111.66 (15) |
| C10A—C11A—H11A | 119.2 | N2B—C17B—H17B | 124.2 |
| C13A—C12A—C11A | 121.89 (15) | C18B—C17B—H17B | 124.2 |
| C13A—C12A—Cl1A | 119.50 (13) | C17B—C18B—N1B | 104.40 (15) |
| C11A—C12A—Cl1A | 118.58 (13) | C17B—C18B—C19B | 127.11 (16) |
| C12A—C13A—C14A | 117.65 (15) | N1B—C18B—C19B | 127.28 (16) |
| C12A—C13A—H13A | 121.2 | O1B—C19B—C18B | 126.31 (19) |
| C14A—C13A—H13A | 121.2 | O1B—C19B—H19B | 116.8 |
| C13A—C14A—C15A | 121.52 (15) | C18B—C19B—H19B | 116.8 |
| C13A—C14A—H14A | 119.2 | C16B—C20B—H20D | 109.5 |
| C15A—C14A—H14A | 119.2 | C16B—C20B—H20E | 109.5 |

| | | | |
|------------------|--------------|-------------------|--------------|
| C14A—C15A—C10A | 120.89 (14) | H20D—C20B—H20E | 109.5 |
| C14A—C15A—N1A | 115.37 (13) | C16B—C20B—H20F | 109.5 |
| C10A—C15A—N1A | 123.74 (13) | H20D—C20B—H20F | 109.5 |
| N2A—C16A—N1A | 111.56 (14) | H20E—C20B—H20F | 109.5 |
| N2A—C16A—C20A | 124.43 (15) | C9BA—C8BA—S1B | 108.8 (2) |
| N1A—C16A—C20A | 123.92 (14) | C9BA—C8BA—H8BA | 109.9 |
| N2A—C17A—C18A | 112.22 (15) | S1B—C8BA—H8BA | 109.9 |
| N2A—C17A—H17A | 123.9 | C9BA—C8BA—H8BB | 109.9 |
| C18A—C17A—H17A | 123.9 | S1B—C8BA—H8BB | 109.9 |
| C17A—C18A—N1A | 104.49 (14) | H8BA—C8BA—H8BB | 108.3 |
| C17A—C18A—C19A | 128.22 (17) | C8BA—C9BA—S2B | 106.3 (2) |
| N1A—C18A—C19A | 127.03 (16) | C8BA—C9BA—H9BA | 110.5 |
| O1A—C19A—C18A | 126.8 (2) | S2B—C9BA—H9BA | 110.5 |
| O1A—C19A—H19A | 116.6 | C8BA—C9BA—H9BB | 110.5 |
| C18A—C19A—H19A | 116.6 | S2B—C9BA—H9BB | 110.5 |
| C16A—C20A—H20A | 109.5 | H9BA—C9BA—H9BB | 108.7 |
| C16A—C20A—H20B | 109.5 | C9BB—C8BB—S1B | 108.3 (14) |
| H20A—C20A—H20B | 109.5 | C9BB—C8BB—H8BC | 110.0 |
| C16A—C20A—H20C | 109.5 | S1B—C8BB—H8BC | 110.0 |
| H20A—C20A—H20C | 109.5 | C9BB—C8BB—H8BD | 110.0 |
| H20B—C20A—H20C | 109.5 | S1B—C8BB—H8BD | 110.0 |
| C8BB—S1B—C7B | 94.2 (6) | H8BC—C8BB—H8BD | 108.4 |
| C8BA—S1B—C7B | 98.91 (10) | C8BB—C9BB—S2B | 105.7 (16) |
| C9BA—S2B—C7B | 95.08 (10) | C8BB—C9BB—H9BC | 110.6 |
| C9BB—S2B—C7B | 99.7 (6) | S2B—C9BB—H9BC | 110.6 |
| C16B—N1B—C18B | 107.07 (13) | C8BB—C9BB—H9BD | 110.6 |
| C16B—N1B—C15B | 125.95 (13) | S2B—C9BB—H9BD | 110.6 |
| C18B—N1B—C15B | 126.91 (13) | H9BC—C9BB—H9BD | 108.7 |
| F1A—C1A—C2A—C3A | 177.86 (16) | F1B—C1B—C6B—C7B | -5.0 (2) |
| C6A—C1A—C2A—C3A | -1.8 (3) | C2B—C1B—C6B—C7B | 175.70 (18) |
| C1A—C2A—C3A—C4A | 0.5 (3) | C4B—C5B—C6B—C1B | -0.4 (2) |
| C2A—C3A—C4A—C5A | 1.0 (3) | C4B—C5B—C6B—C7B | -176.88 (16) |
| C3A—C4A—C5A—C6A | -1.3 (3) | C1B—C6B—C7B—C10B | 67.24 (19) |
| F1A—C1A—C6A—C5A | -178.19 (14) | C5B—C6B—C7B—C10B | -116.45 (16) |
| C2A—C1A—C6A—C5A | 1.4 (2) | C1B—C6B—C7B—S2B | -171.85 (13) |
| F1A—C1A—C6A—C7A | 4.3 (2) | C5B—C6B—C7B—S2B | 4.5 (2) |
| C2A—C1A—C6A—C7A | -176.03 (16) | C1B—C6B—C7B—S1B | -57.09 (17) |
| C4A—C5A—C6A—C1A | 0.1 (2) | C5B—C6B—C7B—S1B | 119.22 (15) |
| C4A—C5A—C6A—C7A | 177.49 (15) | C9BA—S2B—C7B—C6B | 82.91 (18) |
| C1A—C6A—C7A—C10A | -67.81 (18) | C9BB—S2B—C7B—C6B | 106.1 (10) |
| C5A—C6A—C7A—C10A | 114.93 (16) | C9BA—S2B—C7B—C10B | -150.86 (17) |
| C1A—C6A—C7A—S2A | 169.95 (12) | C9BB—S2B—C7B—C10B | -127.7 (10) |
| C5A—C6A—C7A—S2A | -7.31 (19) | C9BA—S2B—C7B—S1B | -31.63 (16) |
| C1A—C6A—C7A—S1A | 54.97 (16) | C9BB—S2B—C7B—S1B | -8.4 (10) |
| C5A—C6A—C7A—S1A | -122.29 (14) | C8BB—S1B—C7B—C6B | -87.5 (7) |
| C9A—S2A—C7A—C6A | -89.25 (13) | C8BA—S1B—C7B—C6B | -111.47 (19) |
| C9A—S2A—C7A—C10A | 144.28 (12) | C8BB—S1B—C7B—C10B | 146.8 (7) |
| C9A—S2A—C7A—S1A | 25.12 (10) | C8BA—S1B—C7B—C10B | 122.8 (2) |
| C8A—S1A—C7A—C6A | 119.73 (12) | C8BB—S1B—C7B—S2B | 32.6 (7) |

supplementary materials

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| C8A—S1A—C7A—C10A | -116.18 (12) | C8BA—S1B—C7B—S2B | 8.64 (19) |
| C8A—S1A—C7A—S2A | -0.96 (11) | C6B—C7B—C10B—C11B | -145.03 (14) |
| C7A—S1A—C8A—C9A | -30.39 (17) | S2B—C7B—C10B—C11B | 89.65 (14) |
| S1A—C8A—C9A—S2A | 51.40 (18) | S1B—C7B—C10B—C11B | -24.74 (17) |
| C7A—S2A—C9A—C8A | -47.47 (16) | C6B—C7B—C10B—C15B | 41.2 (2) |
| C6A—C7A—C10A—C11A | 150.01 (14) | S2B—C7B—C10B—C15B | -84.13 (15) |
| S2A—C7A—C10A—C11A | -83.54 (15) | S1B—C7B—C10B—C15B | 161.48 (12) |
| S1A—C7A—C10A—C11A | 31.36 (17) | C15B—C10B—C11B—C12B | 1.5 (2) |
| C6A—C7A—C10A—C15A | -33.5 (2) | C7B—C10B—C11B—C12B | -172.71 (14) |
| S2A—C7A—C10A—C15A | 92.96 (15) | C10B—C11B—C12B—C13B | -1.7 (3) |
| S1A—C7A—C10A—C15A | -152.14 (13) | C10B—C11B—C12B—C11B | 177.78 (12) |
| C15A—C10A—C11A—C12A | 1.4 (2) | C11B—C12B—C13B—C14B | 0.4 (3) |
| C7A—C10A—C11A—C12A | 178.23 (15) | C11B—C12B—C13B—C14B | -179.10 (14) |
| C10A—C11A—C12A—C13A | -0.3 (3) | C12B—C13B—C14B—C15B | 1.0 (3) |
| C10A—C11A—C12A—C11A | 177.61 (13) | C13B—C14B—C15B—C10B | -1.2 (3) |
| C11A—C12A—C13A—C14A | -0.5 (3) | C13B—C14B—C15B—N1B | 177.64 (16) |
| C11A—C12A—C13A—C14A | -178.33 (14) | C11B—C10B—C15B—C14B | -0.1 (2) |
| C12A—C13A—C14A—C15A | 0.0 (3) | C7B—C10B—C15B—C14B | 173.82 (15) |
| C13A—C14A—C15A—C10A | 1.2 (3) | C11B—C10B—C15B—N1B | -178.85 (13) |
| C13A—C14A—C15A—N1A | -178.58 (15) | C7B—C10B—C15B—N1B | -4.9 (2) |
| C11A—C10A—C15A—C14A | -1.9 (2) | C16B—N1B—C15B—C14B | 81.0 (2) |
| C7A—C10A—C15A—C14A | -178.42 (15) | C18B—N1B—C15B—C14B | -95.36 (19) |
| C11A—C10A—C15A—N1A | 177.91 (14) | C16B—N1B—C15B—C10B | -100.17 (19) |
| C7A—C10A—C15A—N1A | 1.4 (2) | C18B—N1B—C15B—C10B | 83.4 (2) |
| C16A—N1A—C15A—C14A | -75.4 (2) | C17B—N2B—C16B—N1B | 0.81 (19) |
| C18A—N1A—C15A—C14A | 96.07 (19) | C17B—N2B—C16B—C20B | -176.26 (18) |
| C16A—N1A—C15A—C10A | 104.81 (18) | C18B—N1B—C16B—N2B | -1.30 (19) |
| C18A—N1A—C15A—C10A | -83.7 (2) | C15B—N1B—C16B—N2B | -178.28 (14) |
| C17A—N2A—C16A—N1A | -1.26 (19) | C18B—N1B—C16B—C20B | 175.82 (17) |
| C17A—N2A—C16A—C20A | 175.30 (17) | C15B—N1B—C16B—C20B | -1.2 (3) |
| C18A—N1A—C16A—N2A | 2.08 (18) | C16B—N2B—C17B—C18B | 0.0 (2) |
| C15A—N1A—C16A—N2A | 174.99 (14) | N2B—C17B—C18B—N1B | -0.8 (2) |
| C18A—N1A—C16A—C20A | -174.49 (16) | N2B—C17B—C18B—C19B | 167.32 (17) |
| C15A—N1A—C16A—C20A | -1.6 (2) | C16B—N1B—C18B—C17B | 1.20 (17) |
| C16A—N2A—C17A—C18A | -0.1 (2) | C15B—N1B—C18B—C17B | 178.14 (15) |
| N2A—C17A—C18A—N1A | 1.30 (19) | C16B—N1B—C18B—C19B | -166.85 (17) |
| N2A—C17A—C18A—C19A | -173.19 (17) | C15B—N1B—C18B—C19B | 10.1 (3) |
| C16A—N1A—C18A—C17A | -1.97 (17) | C17B—C18B—C19B—O1B | -159.9 (2) |
| C15A—N1A—C18A—C17A | -174.77 (14) | N1B—C18B—C19B—O1B | 5.5 (3) |
| C16A—N1A—C18A—C19A | 172.60 (17) | C8BB—S1B—C8BA—C9BA | -56.2 (13) |
| C15A—N1A—C18A—C19A | -0.2 (3) | C7B—S1B—C8BA—C9BA | 24.1 (4) |
| C17A—C18A—C19A—O1A | 177.3 (2) | S1B—C8BA—C9BA—S2B | -49.0 (4) |
| N1A—C18A—C19A—O1A | 4.0 (3) | C9BB—S2B—C9BA—C8BA | -52.7 (17) |
| F1B—C1B—C2B—C3B | -177.98 (19) | C7B—S2B—C9BA—C8BA | 50.0 (3) |
| C6B—C1B—C2B—C3B | 1.3 (3) | C8BA—S1B—C8BB—C9BB | 50.7 (17) |
| C1B—C2B—C3B—C4B | -0.3 (3) | C7B—S1B—C8BB—C9BB | -51.8 (19) |
| C2B—C3B—C4B—C5B | -1.0 (3) | S1B—C8BB—C9BB—S2B | 50 (2) |
| C3B—C4B—C5B—C6B | 1.4 (3) | C9BA—S2B—C9BB—C8BB | 57.1 (19) |
| F1B—C1B—C6B—C5B | 178.35 (16) | C7B—S2B—C9BB—C8BB | -23.2 (19) |

C2B—C1B—C6B—C5B -0.9 (3)

Hydrogen-bond geometry (Å, °)

Cg1, *Cg2*, *Cg3*, *Cg4* and *Cg5* are the centroids of the C16B—C18B/N1B/N2B, C1B—C6B, C16A—C18A/N1A/N2A, C1A—C6A and C10A—C15A rings, respectively.

| <i>D</i> —H··· <i>A</i> | <i>D</i> —H | H··· <i>A</i> | <i>D</i> ··· <i>A</i> | <i>D</i> —H··· <i>A</i> |
|-------------------------------|-------------|---------------|-----------------------|-------------------------|
| C17B—H17B···O1A | 0.93 | 2.27 | 3.163 (3) | 160 |
| C20A—H20C···F1A | 0.96 | 2.53 | 3.377 (2) | 147 |
| C2A—H2AA···Cg5 ⁱ | 0.93 | 2.81 | 3.727 (2) | 167 |
| C13A—H13A···Cg4 ⁱⁱ | 0.93 | 2.98 | 3.7764 (19) | 144 |
| C13B—H13B···Cg2 ⁱⁱ | 0.93 | 2.93 | 3.716 (2) | 143 |

Symmetry codes: (i) $-x+1, y-1/2, -z+3/2$; (ii) $x, y+1, z$.

supplementary materials

Fig. 1

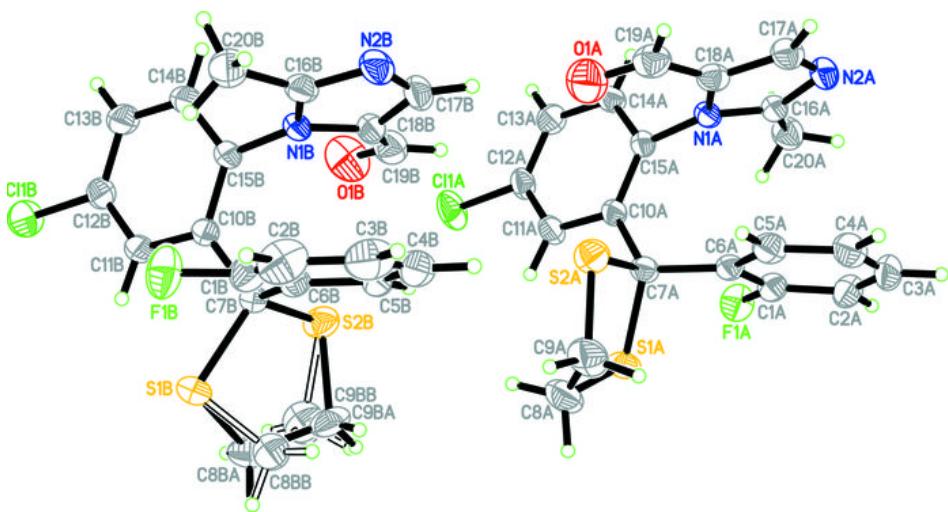


Fig. 2

